



Science	Year 5	Year 6	Year 7	Year 8
Biology	Life exists in a variety of forms and	Living things can be classified according	Cells and tissues	Lungs and gas exchange
	poes through cycles – Animals  Describe the differences in the life cycles of a mammal, an amphibian, an insect and a bird.  Describe the changes as humans develop to old age.	and based on similarities and	living organisms, including how to observe, interpret and record cell structure using a light microscope.  Explain the functions of the cell wall, cell	Explain the structure and functions of the gas exchange system in humans, including adaptations to function.  Describe the mechanism of breathing to move air in and out of the lungs, using a pressure model to explain the movement of gases, including simple measurements
	The human body has a number of systems, each with its own function	animals based on specific characteristics.	mitochondria and chloroplasts.	of lung volume.
	Describe the life process of reproduction in some plants and	The human body has a number of systems, each with its own function	·	Evaluate the impact of exercise, asthma and smoking on the human gas exchange system.
	animals.	Identify and name the main parts of the human circulatory system, and describe	cells.	Drugs and health
		the functions of the heart, blood vessels and blood.	Describe the structural adaptations of	Research the effects of recreational drugs (including substance misuse) on behaviour, health and life processes
		Recognise the impact of diet, exercise, drugs and lifestyle on the way their bodies function.	Describe the hierarchical organisation of	Food and digestion
		Describe the ways in which nutrients and water are transported within	organs, systems to organisms.	Describe the content of a healthy human diet: carbohydrates, lipids (fats and oils), proteins, vitamins, minerals, dietary fibre
		animals, including humans.  Living things exhibit variation and	Muscles and bones  Describe the structure and functions of	and water, and explain why each is needed.
		adaptation and these may lead to evolution	the human skeleton, to include support, protection, movement and making blood cells.	Describe the tissues and organs of the human digestive system, including adaptations to function and how the
		Recognise that living things have changed over time and that fossils provide information about living things	Explain the interaction between skeleton and muscles, including the differences in forces exerted by various muscles	





that inhabited the Earth millions of years	s	
ago.	Consider the function of muscles and	Consider the consequences of
	give examples of antagonistic muscles.	imbalances in the diet, including obesity,
Recognise that living things produce		starvation and deficiency diseases.
offspring of the same kind, but normally		
offspring vary and are not identical to	Animal reproduction	Explain the importance of bacteria in the
their parents.	L	human digestive system.
	Describe reproduction in humans,	
Identify how animals and plants are	including the structure and function of the male and female reproductive	Niversian and Dhatasymthasis
adapted to suit their environment and that adaptation may lead to evolution.	•	Nutrition and Photosynthesis
that adaptation may lead to evolution.	systems.	Describe how plants make carbohydrates
	Explain the stages of the menstrual	in their leaves by photosynthesis and
	cycle.	gaining mineral nutrients and water from
		the soil via their roots.
	Describe the formation of gametes and	
	process of fertilisation.	Give the reactants in, and products of,
		photosynthesis, and a word summary for
	Identify the main stages in gestation and	photosynthesis.
	birth.	
		Explain the dependence of almost all life
	Consider the effect of maternal lifestyle	on Earth on the ability of photosynthetic
	on the foetus through the placenta.	organisms, such as plants and algae, to
	Plant Reproduction	use sunlight in photosynthesis to build organic molecules that are an essential
	Plant Reproduction	energy store and to maintain levels of
	Describe reproduction in plants,	oxygen and carbon dioxide in the
	including flower structure, wind and	atmosphere.
	insect pollination, fertilisation, seed and	
	fruit formation and dispersal, including	
	quantitative	Respiration
	investigation of some dispersal	
	mechanisms.	Describe aerobic and anaerobic
		respiration in living organisms, including
		the breakdown of organic molecules
	Relationships in an Ecosystem	which enables all the other chemical
		processes necessary for life.
	Describe the interdependence of	NA/with a command accommand of the command of the c
	organisms in an ecosystem, including	Write a word summary for aerobic
	food webs and insect pollinated crops.	respiration.





Chemistry  Materials have physical properties which can be investigated and compared  Compare material properties (hardness, solubility, transparency, conductivity and response to magnets).  Know that some materials will dissolve to form a solution, and describe how to recover a substance from a solution.  Use knowledge of solids, liquids and gases of decide how mixtures might be separated, including through filtering, sieving and evaporating.  Demonstrate that dissolving, mixing and changes of state are reversible.  Explain that some hanges result in the formation of new materials and that this kind of change is not usually reversible; (e.g., burning and acid on soda).  The physical properties of materials determine their uses  Materials have physical properties of materials determine their uses  Particles  Atoms and Elements  Aconsider atoms and molecules as particles.  Consider atoms of matter in terms of particles.  Explain diffusion in liquids and gases driven by differences in concentration.  Use knowledge of solids, liquids and gases driven by differences in concentration.  Use chemical symbols and formulae for elements and compounds.  Describe the different states of matter in terms of particles.  Consider gas pressure in terms of particles and apply their understanding to different phenomena including vacuums.  Demonstrate that dissolving, mixing and changes of state are reversible.  Explain diffusion in liquids and gases driven by differences in concentration.  Use chemical symbols and formulae for elements and compounds.  Describe polymers and explain the properties of polymers in terms of their molecular arrangement.  Consider chemical reactions as the rearrangement of atoms.  Consider the importance of the observation of Brownian motion in gases.  Use word equations to represent a chemical reaction.  Pure and impure Substances  The physical properties of materials determine their uses  Give reasons, based on evidence from			Assess the importance of plant reproduction through insect pollination in human food security.  Explain how organisms affect, and are affected by, their environment, including the accumulation of toxic materials.	Assess the process of anaerobic respiration in humans and microorganisms, including fermentation, and a word summary for anaerobic respiration.  Contrast the differences between aerobic
which can be investigated and compared  Compare material properties (hardness, solubility, transparency, conductivity and response to magnets).  Know that some materials will dissolve to form a solution, and describe how to recover a substance from a solution.  Use knowledge of solids, liquids and gases to decide how mixtures might be separated, including through filtering, sieving and evaporating.  Demonstrate that dissolving, mixing and changes of state are reversible. Explain that some changes result in the formation of new materials and that this kind of change is not usually reversible, (e.g., burning and acid on soda).  Which can be investigated and compared to the different states of matter in terms of a terms of arrangements, motion and closeness of particles.  Draw and label a simple atomic model.  Define the key terms atom, element, compound and molecule accurately with specific examples.  Consider gas pressure in terms of particles and apply their understanding to different phenomena including vacuums.  Demonstrate that dissolving, mixing and changes of state are reversible.  Explain diffusion in liquids and gases driven by differences in concentration.  Use chemical symbols and formulae for elements and compounds.  Use chemical symbols and formulae for elements and compounds.  Pure and impure substances of the observation of Brownian motion in gases.  The physical properties of materials determine their uses  Give reasons, based on evidence from  Describe the varying physical and chemical reactionts.  Describe the varying physical and chemical reactions as the rearrangement.				reactants, the products formed and the implications for the organism.
Compared  Compare material properties (hardness, solubility, transparency, conductivity and response to magnets).  Know that some materials will dissolve to form a solution, and describe how to recover a substance from a solution.  Use knowledge of solids, liquids and gases to decide how mixtures might be separated, including through filtering, sieving and evaporating.  Demonstrate that dissolving, mixing and changes of state are reversible.  Explain diffusion in liquids and gases to decide how mixtures might be separated, including through filtering, sieving and evaporating.  Demonstrate that dissolving, mixing and changes of state are reversible.  Explain diffusion in liquids and gases in concentration.  Use knowledge of solids, liquids and gases to decide how mixtures might be separated, including through filtering, sieving and evaporating.  Demonstrate that dissolving, mixing and changes of state in terms of particles.  Appreciate the importance of the observation of Brownian motion in gases.  Consider chemical reactions as the rearrangement.  Consider chemical reactions as the rearrangement of atoms.  Consider chemical reactions as the rearrangement of atoms.  Consider chemical reactions as the rearrangement of atoms.  Consider chemical reaction as the rearrangement of atoms.  Consider chemical symbols and formulae for elements and compounds.  Use chemical symbols and formulae for elements and compounds and including the proper	Chemistry		Particles	Atoms and Elements
terms of arrangements, motion and closeness of particles.  Explain changes of state in terms of energy.  Explain changes of state in terms of energy.  Explain diffusion in liquids and gases driven by differences in concentration.  Use knowledge of solids, liquids and gases to decide how mixtures might be separated, including through filtering, sieving and evaporating.  Demonstrate that dissolving, mixing and changes of state are reversible. Explain that some changes result in the formation of new materials and that this kind of change is not usually reversible, (e.g., burning and acid on soda).  The physical properties of materials Give reasons, based on evidence from  The physical properties of materials  The periodic table and reactivity  terms of arrangements, motion and closeness of particles.  Draw and label a simple atomic model.  Define the key terms atom, element, compound and molecule accurately with specific examples.  Use chemical symbols and formulae for elements and compounds.  Use chemical symbols and formulae for elements and compounds.  Describe polymers and explain the properties of polymers in terms of their molecular arrangement.  Consider the importance of the observation of Brownian motion in gases.  The physical properties of materials  Give reasons, based on evidence from				
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determine their uses       Recognise particle diagrams elements, compounds and mixtures.       Describe the varying physical and chemical		The physical properties of materials	Pure and impure substances	The periodic table and reactivity
Give reasons, based on evidence from compounds and mixtures.  Describe the varying physical and chemical			Recognice particle diagrams elements	The periodic table and reactivity
Give reasons, based on evidence from chemical		determine their uses		Describe the varying physical and
		Give reasons, based on evidence from	compounds and mixtures.	
I ICOMPARATIVE AND TAIR TESTS. FOR THE I I INTONEPTIES OF DIFFERENT ELEMENTS.		comparative and fair tests, for the		properties of different elements.





Use melting and boiling points to	
determine a pure or impure substance.	Describe the principles underpinning the Mendeleev Periodic Table
Use the key terms dissolve, solute,	
solvent and solution.	Describe the Periodic Table: periods and groups; metals and non-metals.
Undertake simple techniques for	
separating mixtures: filtration,	Compare the physical and chemical
·	properties of metals and non-metals
<b>0</b> 1 ,	Explain how patterns in reactions can be
	predicted with reference to the Periodic Table.
Chemical Reactions	
	Research the properties of metals and
	non-metals.
involve the rearrangement of atoms.	Fouth and Atmosphane
Define acids and alkalis in terms of	Earth and Atmosphere
neutralisation reactions.	Describe the composition of the Earth.
Use the pH scale for measuring	Recall the structure of the Earth.
acidity/alkalinity; and indicators.	
Investigate the reactions of saids with	Describe the rock cycle and the formation of igneous, sedimentary and
metals to produce a salt plus hydrogen.	metamorphic rocks.
Investigate the reactions of acids with	
alkalis to produce a salt plus water.	Consider Earth as a source of limited resources and the efficacy of recycling.
	Describe the carbon cycle and
	anthropogenic contribution.
	Recall the composition of the
	atmosphere
	Consider the production of carbon
	dioxide by human activity and the impact on climate.
	determine a pure or impure substance.  Use the key terms dissolve, solute, solvent and solution.  Undertake simple techniques for separating mixtures: filtration, evaporation, distillation and chromatography.  Chemical Reactions  Understand that chemical reactions involve the rearrangement of atoms.  Define acids and alkalis in terms of neutralisation reactions.  Use the pH scale for measuring acidity/alkalinity; and indicators.  Investigate the reactions of acids with metals to produce a salt plus hydrogen.





Physics	There are contact and non-contact	Light & sound can be reflected &	Forces and motion	Forces
PHYSICS	forces; these affect the motion of	absorbed and enable us to see & hear	l orces and motion	Describe forces: associated with
	objects		Describe forces as pushes or pulls, arising	
	Explain that unsupported objects fall	Recognise that light appears to travel in		squashing – springs; friction between
	1 ' ' ' ' ' ' ' ' ' ' ' ' ' ' ' ' ' ' '	straight lines.	objects.	surfaces, pushing things out of the way
	of gravity acting between the Earth and		Use force arrows in diagrams, adding	resistance of air and water.
	the falling object.	Use the idea that light travels in straight		
		lines to explain that objects are seen	unbalanced forces.	Define forces as measured in newtons
	Identify the effects of air resistance,	because they give out or reflect light		measurements of stretch or compressi
	water resistance and friction, that act	into the eye.	Study speed and the quantitative	as the force applied is changed.
	between moving surfaces.	·	relationship between average speed,	
		Explain that we see things because light	distance and time (speed = distance ÷	Investigate force-extension linear
	Recognise that some mechanisms,	travels from light sources to our eyes or	time).	relation; Hooke's Law.
	including levers, pulleys and gears, allow	from light sources to objects and to		
	a smaller force to have a	our eyes.	Represent a journey on a distance-time	Consider work done and energy chang
	greater effect.		graph.	on deformation.
		Use the idea that light travels in straight		
	Day, night, month, seasons& years are	lines to explain why shadows have the		
	caused by the position and movement	same shape as the objects		Energy Transfers
	of the Earth	that cast them.	Describe the forces being needed to	
			cause objects to stop or start moving, or	Define the term energy as the ability to
	Describe the movement of the Earth,	Electricity can make circuits work and	to change their speed or direction of	do work.
	and other planets, relative to the Sun in	can be controlled to perform useful	motion (qualitative only).	
	the solar system.	functions		
			Describe opposing forces and	Describe heating and thermal
	Describe the movement of the Moon	Associate the brightness of a lamp or the		equilibrium: temperature difference
	relative to the Earth.	volume of a buzzer with the number and		between two objects leading to energ
		voltage of cells used in a circuit.	compressed surface.	transfer from the hotter to the cooler
	Describe the Sun, Earth and Moon as	Compare and give reasons for variations		one through contact (conduction) or
	approximately spherical bodies.	in how components function, including	Investigate non-contact forces: gravity	radiation; such transfers tending to
	Use the idea of the Earth's rotation to	-	forces acting at a distance on Earth and	reduce the temperature difference.
	-  -  -  -  -  -  -  -  -  -  -  -  -	and switches.	in space, forces between magnets and	
	movement of the sun across		forces due to static electricity.	Assess the use of insulators to minimis
	sky.	Use recognised symbols when		heat transfer.
		representing a simple circuit in a	Energy	
		diagram.	Consider energy as a quantity that can	
			be quantified and calculated.	Electricity and Electromagnetism





Explain why the total energy has the Describe electric currents, measured i
same value before and after a change amperes, in circuits, series and paralle
(conservation of energy) circuits, currents add where branches
meet and current as flow of charge.  Compare the starting and the final
conditions of a system and describing Investigate potential difference,
increases and decreases in the amounts measured in volts, battery and bulb
of energy associated with movements, ratings; resistance, measured in ohms
temperatures (not covered in detail: the ratio of potential difference (p.d.)
changes in positions in a field, in elastic current.
distortions and in chemical
compositions). Investigate differences in resistance
Explain changes with temperature in the between conducting and insulating
motion and spacing of particles. components (quantitative).
Compare energy values of different Describe electrostatic forces as the
foods (from labels) (kJ). separation of positive or negative
charges when objects are rubbed
Compare power ratings of appliances in together: transfer of electrons.
watts (W, kW).
Explain the idea of electric field, force
Calculate and compare amounts of acting across the space between objection
energy transferred (J, kJ, kW hour). not in contact.
Assess domestic fuel bills, fuel use and Consider magnetic poles, attraction a
costs. repulsion.
Compare fuels and consider renewable
and non-renewable energy resources. Plot magnetic fields with compass,
representation by field lines (HW
project).
Waves Study Earth's magnetism, compass a
navigation (HW project).
Explain that sound is produced by
vibrations of objects, in loud speakers, Investigate the magnetic effect of a
detected by their effects on microphone current, electromagnets, D.C. motors
diaphragm and the ear drum; sound (principles only).
waves are longitudinal.
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Describe the auditory range of humans and animals.





			Investigate the transmission of light through materials: absorption, diffuse scattering and specular reflection.  Use the ray model to explain imaging in mirrors, the pinhole camera, the refraction of light and action of convex lens in focusing (qualitative); the eye.  Consider light as transferring energy from source to absorber leading to chemical and electrical effects; photosensitive material in the retina and in cameras.  Describe colours using different frequencies of light, white light and prisms (qualitative only); differential colour effects in absorption and diffuse reflection.	
Working	Planning investigations	Planning investigations	Scientific Attitudes	Scientific Attitudes
Scientifically	Pupils can plan an enquiry.	Pupils can ask questions.	Identify variables and describe how to control them. Begin to evaluate	Identify variables and describe how to control them. Begin to evaluate accuracy,
	Pupils can begin to identify variables.	Pupils can plan an enquiry.	accuracy, precision, repeatability and reproducibility.	precision, repeatability and reproducibility.
		Pupils can identify		
	Conducting experiments	Conducting experiments	Understand that scientific methods and	Understand that scientific methods and
	Dunile can use equipment to take	Dunile can use equipment to take	theories develop as earlier explanations are modified to take account of new	theories develop as earlier explanations are modified to take account of new
	Pupils can use equipment to take	Pupils can use equipment to take	evidence and ideas, together with the	evidence and ideas, together with the
	measurement s.	measurements.	_	=
	Pupils explore how to improve the	Pupils explore how to improve the	importance of publishing results and peer review.	importance of publishing results and peer review.
	quality of data.	quality of data.	peer review.	peer review.
		quanty of data.		





Pupils understand the role of repeat	Pupils understand the role of repeat		Experimental Skills and Investigations
readings.	readings.	Experimental Skills and Investigations	
			Ask questions and develop a line of
Recording evidence	Recording evidence		enquiry based on observations of the real
			world, alongside prior knowledge and
Pupils record work with diagrams and	Pupils record work with diagrams and	, , ,	experience.
label them.	label them.	and experience.	
			Make predictions using scientific
Pupils can display data using labelled	Pupils can display data using labelled		knowledge and understanding.
diagrams, keys, tables and bar charts.	diagrams, keys, tables and bar charts.	knowledge and understanding.	Calast when and anymy and the mast
Pupils can display data using line	Pupils can display data using line		Select, plan and carry out the most appropriate types of scientific enquiries
graphs.	graphs.		to test predictions, including identifying
βιαριίο.	graphs.		independent, dependent and control
Reporting findings	Reporting findings		variables, where appropriate.
reporting internet	neporting manage	variables, where appropriate.	variables, where appropriate.
Pupils process findings to develop	Pupils process findings to develop		Use appropriate techniques, apparatus,
conclusions and identify causal	conclusions and identify causal		and materials during fieldwork and
relationships.	relationships.		laboratory work, paying attention to
·		laboratory work, paying attention to	health and safety.
Pupils use displays and presentations to	Pupils use displays and presentations to	health and safety.	
report on findings.	report on findings.		Make and record observations and
		Make and record observations and	measurements using a range of methods
Pupils explain confidence in findings.	Pupils explain confidence in findings	measurements using a range of methods	
		for different investigations; and evaluate	
Conclusions and predictions	Conclusions and predictions		possible
		possible improvements.	improvements.
Pupils can analyse data.	Pupils can analyse data.	A male consoling to about our	A male, and maline to the single
Dunils can draw canalysians	Dunils can draw conclusions	Apply sampling techniques.	Apply sampling techniques.
Pupils can draw conclusions.	Pupils can draw conclusions.	Analysis and evaluation	Analysis and evaluation
Suggest further comparative or fair	Pupils can develop investigation further.	Allalysis allu evaluation	Alialysis and evaluation
tests.	apilo cuit develop investigation further.	Apply mathematical concepts and	Apply mathematical concepts and
10010		calculate results.	calculate results.
		Present observations and data using	Present observations and data using
			appropriate methods, including tables
		and graphs.	and graphs.





			including identifying patterns and using observations, measurements and data to draw conclusions.  Present reasoned explanations, including explaining data in relation to predictions and hypotheses.  Evaluate data, showing awareness of potential sources of random and systematic error.  Identify further questions arising from	Interpret observations and data, including identifying patterns and using observations, measurements and data to draw conclusions.  Present reasoned explanations, including explaining data in relation to predictions and hypotheses.  Evaluate data, showing awareness of potential sources of random and systematic error.  Identify further questions arising from their results.
			Measurement	Measurement
			Understand and use SI units and chemical nomenclature.	Understand and use SI units and chemical nomenclature.
				Use and derive simple equations and carry out appropriate calculations.
			,	Undertake basic data analysis including simple statistical techniques.
Demonstrating	Explain the reasons for changes as	Justify the classification of unusual types	Explain how specialised cells are adapted	Explain how the gas exchange and
Greater Depth in	,	1	to carry out their function.	digestive systems are adapted to carry out their function.
	animals and plants		Explain how the skeletal and muscular system are adapted to carry out their	Explain how leaves are adapted to carry out photosynthesis
	make it useful.  Explain changes of state in terms of	organisms enable survival.	occur during puberty.	Explain the effects of exercise on aerobic and anaerobic respiration
	= =	the evolution of a species	Explain how a plants reproductive parts, including seeds are adapted to carry out their function.	





Explain the effect of surface area on a	Explain how the interaction of light with	Explain how changes in population may	Name compounds from chemical
resistance or water resistance	different media may affect what is seen.	,	formulae
	·		
Evaluate evidence for the force of		1	Explain the properties of sedimentary,
gravity, the shape of Earth and the		and gases in terms of particle energy and	metamorphic and igneous rock.
Heliocentric model		particle arrangement	Explain factors affecting drag forces in
Correctly identify variables and sugges	current and voltage. t		terms of particles
improvements to investigation		have sharp melting points	
methodology			Compare methods thermal energy
		l	transfer (conduction, convection and
		terms of particles	radiation)
		Describe concentrated and dilute acids in	Explain how charged particles cause
			static electricity in terms of repulsion or
			attraction
		Name salts formed from neutralisation	
		1	Explain the factors affecting the strength
		describe these reactions	of an electromagnet
		Explain the meaning of relative motion:	
		trains and cars passing one another.	
		Explain how an object's mass affects the	
		force of gravity.	
		Explain energy transfer from one source	
		to another	
		Suggest how the efficiency of an	
		appliance may be improved, using ideas	
		of dissipation.	
		Explain the relationship between	
		frequency, pitch, loudness and amplitude	